# Salvinorin, a New trans-Neoclerodane Diterpene from Salvia divinorum (Labiatae) 

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#### Abstract

Salvinorin, isolated from Salvia divinorum, has been shown by spectroscopic and $X$-ray-crystallographic methods to be a trans-neoclerodane diterpene of structure (1). Crystals of compound (1) are orthorhombic, space group $P 2,2,2$, with $a=6.368(2), b=11.338(3), c=30.710(6) A$, and $Z=4$. The structure was refined by leastsquares to $R 0.052$ and $R^{\prime} 0.056$.


The essential oils produced by certain members of the widespread genus Salvia (Labiatae) are used extensively in the food and cosmetic industries. Examples are Dalmatian sage oil from $S$. officinalis (used to flavour certain foods) and Clary sage oil from S. sclarea (used in perfumery). ${ }^{1}$ S. divinorum (' hojas de la Pastora', possibly identical with 'pipilzintzintli') is a relatively rare plant that is used by the Mazatec Indians of Mexico in their divination rites, ${ }^{2}$ but no previous chemical studies have been reported for it. However, various biand tri-cyclic diterpenes have been isolated from other Salvia species. ${ }^{3}$ Extraction of the leaves of S. divinorum has now yielded a novel bicyclic diterpene, salvinorin (1), $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{O}_{8}$, whose structural eludication forms the subject of this paper.

Although the i.r. spectrum $\left(\mathrm{CHCl}_{3}\right)$, of salvinorin (1) showed only one peak in the carbonyl region ( $v_{\text {max. }} 1735$ $\left.\mathrm{cm}^{-1}\right)$, the ${ }^{13} \mathrm{C}$ n.m.r. spectrum $\left(\mathrm{CDCl}_{3} ; \delta /\right.$ p.p.m. $)$ revealed carbons due to four carbonyl groups: one of the ketone type (singlet at 202.04) and three of the ester type (singlets at 171.57, 171.15, and 169.94). Other salient features in the ${ }^{13} \mathrm{C}$ n.m.r. spectrum of compound (1) included absorptions due to a $\beta$-substituted furan (singlet at 125.25 and doublets at $143.66,139.46$, and 108.41), four methyl carbons (quartets at $51.90,20.56$, 16.36 , and 15.19 ), and two methine carbons bearing



Figure An ORTEP stereoscopic drawing of salvinorin (1)

(1)

(2) $R=H$
(3) $R=O A C$
oxygen (doublets at 75.03 and 72.00 , these are assigned to C-2 and C-12, respectively). There were also absorptions due to three methine carbons $\alpha$ to carbonyl groups (doublets at $63.90,53.47$, and 51.26 ), four unassigned methylene carbons (triplets at 43.23, 38.08, 30.75, and 18.11), and two quaternary carbons (singlets at 42.06 and 35.41). The ${ }^{1} \mathrm{H}$ n.m.r. spectrum $\left(\mathrm{CDCl}_{3}\right)$ had absorptions due to two tertiary methyl groups (singlets at $\delta_{\mathrm{H}} 1.11$ and 1.45), a methyl ester (singlet at 3.74) and the $\beta$-substituted furan ( 1 H -multiplet at 6.38 and 2 H -multiplet

at 7.41). Absorption due to the acetate appeared at $\delta_{\mathrm{H}} 2.16$; that the acetate was a secondary one was evident from the presence of a one-proton triplet $\left(\delta_{H} 5.14\right.$, $J 10 \mathrm{~Hz}$ ). A one-proton doublet of doublets ( $\delta_{\mathrm{H}} 5.51$, $J 12$ and 6 Hz ) is assigned to the $12-\mathrm{H}$.

Final proof of the stereostructure of salvinorin (1) was obtained from a single-crystal $X$-ray analysis using direct methods. ${ }^{4}$ Details of the $X$-ray analysis are given in the Experimental section, and listings of final atomic

## TABLE 1

Final atomic parameters for salvinorin (1) (standard deviations in parentheses)

| Atom | $x / a$ | $y / b$ | $z / c$ | $B$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O}(1)$ | $0.2814(6)$ | 0.479 6(3) | $0.3695(1)$ | - |
| $\mathrm{O}(2)$ | 0.199 0(6) | 0.542 l (3) | 0.288 2(1) | - |
| $\mathrm{O}(3)$ | $0.3462(9)$ | $0.3652(4)$ | $0.2796(2)$ | - |
| $\mathrm{O}(4)$ | 0.7603 (9) | $0.9511(4)$ | 0.325 9(1) | - |
| $\mathrm{O}(5)$ | $0.4092(11)$ | 0.961 3(4) | 0.323 9(2) | - |
| $\mathrm{O}(6)$ | 0.9358 (6) | $0.4059(3)$ | 0.478 2(1) | - |
| $\mathrm{O}(7)$ | 1.020 6(6) | 0.574 6(3) | 0.5078 (1) | - |
| $\mathrm{O}(8)$ | $1.0231(8)$ | 0.039 6(3) | 0.430 6(1) | - |
| $\mathrm{O}(11)$ | $0.2551(39)$ | $0.1312(20)$ | 0.2711 (7) | 14.0 (11) |
| $\mathrm{C}(1)$ | 0.413 3(8) | $0.5427(5)$ | $0.3531(2)$ |  |
| $\mathrm{C}(2)$ | 0.3930 (9) | $0.5842(5)$ | $0.3063(2)$ | - |
| C(3) | $0.3895(10)$ | $0.7186(5)$ | $0.3045(2)$ | - |
| $\mathrm{C}(4)$ | 0.574 3(10) | $0.7718(5)$ | 0.329 6(2) | - |
| C(5) | 0.5800 (8) | 0.7280 (4) | 0.378 5(2) | - |
| C(6) | 0.775 7(8) | 0.7793 (5) | $0.4012(2)$ | - |
| C(7) | $0.8308(9)$ | 0.7199 (5) | 0.4445 (2) | - |
| C(8) | 0.858 8(8) | $0.5879(5)$ | $0.4377(1)$ | - |
| $\mathrm{C}(9)$ | $0.6592(8)$ | 0.527 4(4) | 0.419 6(1) | - |
| $\mathrm{C}(10)$ | $0.6025(7)$ | 0.590 6(4) | $0.3765(1)$ | - |
| C(11) | $0.7279(8)$ | $0.4008(5)$ | $0.4093(2)$ | - |
| C(12) | 0.804 8(9) | $0.3335(5)$ | $0.4494(2)$ | - |
| C(13) | 0.9290 (9) | 0.2275 (5) | 0.4390 (2) | - |
| C(14) | 1.125 7(10) | $0.2211(5)$ | 0.417 6(2) | - |
| C(15) | 1.1771 (12) | $0.1093(6)$ | $0.4140(2)$ | - |
| C(16) | 0.8741 (10) | 0.1149 (5) | 0.445 6(2) | - |
| C(17) | 0.9389 (8) | $0.5260(5)$ | 0.477 6(2) | - |
| $\mathrm{C}(18)$ | $0.5667(15)$ | $0.9042(6)$ | $0.3259(2)$ | - |
| $\mathrm{C}(19)$ | $0.3798(9)$ | $0.7703(5)$ | 0.402 2(2) | - |
| $\mathrm{C}(20)$ | $0.4859(8)$ | 0.526 7(5) | 0.454 8(2) | - |
| C(21) | $0.1945(12)$ | $0.4259(7)$ | 0.2790 (2) | - |
| $\mathrm{C}(22)$ | -0.025 3(12) | 0.3873 (7) | $0.2682(2)$ | - |
| $\mathrm{C}(23)$ | $0.7679(18)$ | 1.079 3(6) | 0.324 8(2) |  |
| $\mathrm{H}(2)$ | 0.513 | 0.554 | 0.289 | 4.5 |
| $\mathrm{H}(3) \mathrm{A}$ | 0.254 | 0.745 | 0.318 | 5.0 |
| $\mathrm{H}(3) \mathrm{B}$ | 0.397 | 0.742 | 0.274 | 5.0 |
| $\mathrm{H}(4)$ | 0.705 | 0.744 | 0.316 | 4.0 |
| $\mathrm{H}(6) \mathrm{A}$ | 0.899 | 0.768 | 0.381 | 4.0 |
| $\mathrm{H}(6) \mathrm{B}$ | 0.753 | 0.864 | 0.406 | 4.0 |
| H(7)A | 0.716 | 0.735 | 0.465 | 4.0 |
| $\mathrm{H}(7) \mathrm{B}$ | 0.966 | 0.755 | 0.455 | 4.0 |
| $\mathrm{H}(8)$ | 0.971 | 0.578 | 0.415 | 3.5 |
| $\mathrm{H}(10)$ | 0.725 | 0.577 | 0.357 | 3.0 |
| H(llA) | 0.605 | 0.357 | 0.397 | 4.0 |
| H(11)B | 0.844 | 0.403 | 0.387 | 4.0 |
| $\mathrm{H}(12)$ | 0.681 | 0.308 | 0.467 | 4.0 |
| H(14) | 1.208 | 0.290 | 0.407 | 6.0 |
| $\mathrm{H}(15)$ | 1.309 | 0.078 | 0.401 | 7.0 |
| $\mathrm{H}(16)$ | 0.738 | 0.090 | 0.460 | 5.0 |
| $\mathrm{H}(19) \mathrm{A}$ | 0.383 | 0.743 | 0.433 | 5.0 |
| H(19) B | 0.253 | 0.738 | 0.388 | 5.0 |
| $\mathrm{H}(19) \mathrm{C}$ | 0.376 | 0.859 | 0.402 | 5.0 |
| $\mathrm{H}(20) \mathrm{A}$ | 0.359 | 0.485 | 0.443 | 4.0 |
| $\mathrm{H}(20) \mathrm{B}$ | 0.446 | 0.610 | 0.462 | 4.0 |
| $\mathrm{H}(20) \mathrm{C}$ | 0.538 | 0.485 | 0.481 | 4.0 |
| $\mathrm{H}(22) \mathrm{A}$ | -0.095 | 0.360 | 0.295 | 8.0 |
| $\mathrm{H}(22) \mathrm{B}$ | -0.013 | 0.318 | 0.247 | 8.0 |
| $\mathrm{H}(22) \mathrm{C}$ | -0.102 | 0.452 | 0.254 | 8.0 |
| $\mathrm{H}(23) \mathrm{A}$ | 0.915 | 1.109 | 0.326 | 12.0 |
| $\mathrm{H}(23) \mathrm{B}$ | 0.692 | 1.113 | 0.352 | 12.0 |
| $\mathrm{H}(23) \mathrm{C}$ | 0.694 | 1.111 | 0.298 | 12.0 |

Table 2
Bond lengths ( $\AA$ ) in salvinorin (1) (standard deviations in parentheses)

| $\mathrm{O}(1)-\mathrm{C}(1)$ | $1.213(6)$ | $\mathrm{C}(4)-\mathrm{C}(18)$ | $1.507(8)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O}(2)-\mathrm{C}(2)$ | $1.437(7)$ | $\mathrm{C}(5)-\mathrm{C}(6)$ | $1.541(7)$ |
| $\mathrm{O}(2)-\mathrm{C}(21)$ | $1.348(9)$ | $\mathrm{C}(5)-\mathrm{C}(10)$ | $1.566(7)$ |
| $\mathrm{O}(3)-\mathrm{C}(21)$ | $1.186(9)$ | $\mathrm{C}(5)-\mathrm{C}(19)$ | $1.544(7)$ |
| $\mathrm{O}(4)-\mathrm{C}(18)$ | $1.343(11)$ | $\mathrm{C}(6)-\mathrm{C}(7)$ | $1.531(8)$ |
| $\mathrm{O}(4)-\mathrm{C}(23)$ | $1.455(8)$ | $\mathrm{C}(7)-\mathrm{C}(8)$ | $1.522(7)$ |
| $\mathrm{O}(5)-\mathrm{C}(18)$ | $1.195(11)$ | $\mathrm{C}(8)-\mathrm{C}(9)$ | $1.548(7)$ |
| $\mathrm{O}(6)-\mathrm{C}(12)$ | $1.468(6)$ | $\mathrm{C}(8)-\mathrm{C}(17)$ | $1.500(7)$ |
| $\mathrm{O}(6)-\mathrm{C}(17)$ | $1.362(7)$ | $\mathrm{C}(9)-\mathrm{C}(10)$ | $1.548(6)$ |
| $\mathrm{O}(7)-\mathrm{C}(17)$ | $1.199(6)$ | $\mathrm{C}(9)-\mathrm{C}(11)$ | $1.534(7)$ |
| $\mathrm{O}(8)-\mathrm{C}(15)$ | $1.358(8)$ | $\mathrm{C}(9)-\mathrm{C}(20)$ | $1.544(7)$ |
| $\mathrm{O}(8)-\mathrm{C}(16)$ | $1.357(8)$ | $\mathrm{C}(11)-\mathrm{C}(12)$ | $1.528(7)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.518(7)$ | $\mathrm{C}(12)-\mathrm{C}(13)$ | $1.473(8)$ |
| $\mathrm{C}(1)-\mathrm{C}(10)$ | $1.504(7)$ | $\mathrm{C}(13)-\mathrm{C}(14)$ | $1.417(8)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.524(8)$ | $\mathrm{C}(13)-\mathrm{C}(16)$ | $1.340(8)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | $1.531(8)$ | $\mathrm{C}(14)-\mathrm{C}(15)$ | $1.313(9)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.582(7)$ | $\mathrm{C}(21)-\mathrm{C}(22)$ | $1.504(11)$ |

Table 3
Bond angles $\left({ }^{\circ}\right)$ in salvinorin ( 1 ) (standard deviations in parentheses)

| $\mathrm{C}(2)-\mathrm{O}(2)-\mathrm{C}(21)$ | 115.1(5) |
| :---: | :---: |
| $\mathrm{C}(18)-\mathrm{O}(4)-\mathrm{C}(23)$ | 115.2(7) |
| $\mathrm{C}(12)-\mathrm{O}(6)-\mathrm{C}(17)$ | 123.9(4) |
| $\mathrm{C}(15)-\mathrm{O}(8)-\mathrm{C}(16)$ | 105.5(5) |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(2)$ | 121.1(5) |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(10)$ | 124.7(4) |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(10)$ | 114.1(4) |
| $\mathrm{O}(2)-\mathrm{C}(2)-\mathrm{C}(1)$ | 109.7(4) |
| $\mathrm{O}(2)-\mathrm{C}(2)-\mathrm{C}(3)$ | 107.8(4) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 110.3(4) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | $111.3(5)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 111.9(4) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(18)$ | 109.3(5) |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{C}(18)$ | 112.7(4) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | 109.2(4) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(10)$ | 106.1(4) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(19)$ | 109.3(4) |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(10)$ | 108.6(4) |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(19)$ | 109.7(4) |
| $\mathrm{C}(10)-\mathrm{C}(5)-\mathrm{C}(19)$ | 113.8(4) |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | 114.3(4) |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | 109.9(4) |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | 112.9(4) |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(17)$ | 112.9(4) |
| $\mathrm{C}(9)-\mathrm{C}(8)-\mathrm{C}(17)$ | 111.3(4) |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | 107.0(4) |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(11)$ | 104.8(4) |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(20)$ | 109.8(4) |
| $\mathrm{C}(10)-\mathrm{C}(9)-\mathrm{C}(11)$ | 108.9(4) |
| $\mathrm{C}(10)-\mathrm{C}(9)-\mathrm{C}(20)$ | 115.7(4) |
| $\mathrm{C}(11)-\mathrm{C}(9)-\mathrm{C}(20)$ | 110.1(4) |
| $\mathrm{C}(1)-\mathrm{C}(10)-\mathrm{C}(5)$ | 107.7(4) |
| $\mathrm{C}(1)-\mathrm{C}(10)-\mathrm{C}(9)$ | 115.3(4) |
| $\mathrm{C}(5)-\mathrm{C}(10)-\mathrm{C}(9)$ | 116.6(4) |
| $\mathrm{C}(9)-\mathrm{C}(11)-\mathrm{C}(12)$ | 113.1(4) |
| $\mathrm{O}(6)-\mathrm{C}(12)-\mathrm{C}(11)$ | 112.9(4) |
| $\mathrm{O}(6)-\mathrm{C}(12)-\mathrm{C}(13)$ | 106.4(4) |
| $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(13)$ | 113.8(4) |
| $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(14)$ | 128.2(5) |
| $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(16)$ | 127.2(5) |
| $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(16)$ | 104.5(5) |
| $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)$ | 108.0(6) |
| $\mathrm{O}(8)-\mathrm{C}(15)-\mathrm{C}(14)$ | 110.5(6) |
| $\mathrm{O}(8)-\mathrm{C}(16)-\mathrm{C}(13)$ | 111.4(5) |
| $\mathrm{O}(6)-\mathrm{C}(17)-\mathrm{O}(7)$ | 117.0(5) |
| $\mathrm{O}(6)-\mathrm{C}(17)-\mathrm{C}(8)$ | 118.4(4) |
| $\mathrm{O}(7)-\mathrm{C}(17)-\mathrm{C}(8)$ | 124.4(5) |
| $\mathrm{O}(4)-\mathrm{C}(18)-\mathrm{O}(5)$ | 123.8(6) |
| $\mathrm{O}(4)-\mathrm{C}(18)-\mathrm{C}(4)$ | 111.4(7) |
| $\mathrm{O}(5)-\mathrm{C}(18)-\mathrm{C}(4)$ | 124.8(8) |
| $\mathrm{O}(2)-\mathrm{C}(21)-\mathrm{O}(3)$ | 123.1(7) |
| $\mathrm{O}(2)-\mathrm{C}(21)-\mathrm{C}(22)$ | 110.5(6) |
| $\mathrm{O}(3)-\mathrm{C}(21)-\mathrm{C}(22)$ | 126.3(7) |

parameters, bond lengths, bond angles, and torsion angles are given in Tables 1-4. An ORTEP stereoscopic drawing of compound (1), as determined from the $X$-raycrystallographic analysis, is displayed in the Figure. This figure also represents the absolute stereochemistry of salvinorin, which was deduced from the negative c.d. curve ( $294 \mathrm{~nm}, \varepsilon-5600$ in dioxan) due the keto-group at C-1, in accord with that reported for isofructicolone. ${ }^{5}$

Table 4
Torsion angles $\left({ }^{\circ}\right)$ in salvinorin (1) (standard deviations in parentheses)


Salvinorin (1) thus belongs to the neoclerodane class of diterpenes, a group of compounds that has attracted considerable interest because of problems associated with their stereochemistry ${ }^{6}$ and because of the diverse biological activities shown by some members (e.g. insect antifeedant, antitumour, and antifungal properties). ${ }^{7}$ Except for differences in the substituents and the stereochemistry at C-8 and C-12, salvinorin (1) is structurally similar to salviarin (2) ${ }^{3}$ and splendidin (3), ${ }^{8}$ compounds which were recently isolated from S. splendens by Hanson and his collaborators.

## EXPERIMENTAL

The m.p. was determined in a capillary tube. I.r. and n.m.r. spectra were determined in chloroform and deuteriochloroform, respectively. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ n.m.r. spectra were determined at 200 and 50.8 MHz , respectively. Chemical shifts are expressed in p.p.m. downfield from tetramethylsilane as internal reference, with coupling constants $(J)$ in Hz . The mass spectrum was recorded at $70 \mathrm{eV} ; m / z$ values are given with relative intensities (\%) in parentheses. Thin-layer chromatography (t.l.c.) was

* 'Tonsil' is a commercially available bentonitic earth with the following composition: $\mathrm{SiO}_{2}(72.5 \%), \mathrm{Al}_{2} \mathrm{O}_{3}(13 \%), \mathrm{Fe}_{2} \mathrm{O}_{3}$ ( $5 \%$ ), $\mathrm{MgO}(1.5 \%), \mathrm{CaO}(7.2 \%)$, and $\mathrm{H}_{2} \mathrm{O}(8.5 \%)$, and has $\mathrm{pH}^{3}$ 3.
$\dagger$ For details see Notice to Authors No. 7 in J. Chem. Soc., Perkin Trans. 1, 1981, Index issue.
performed on silica ( $\mathrm{PF}_{254}$, Merck) plates and spots were made visible by spraying with $10 \%$ phosphomolybdic acid in propan-2-ol, followed by heating. Column chromatography was carried out using 'Tonsil' as adsorbent.

Isolation of Salvinorin (1).--Dried, milled leaves (200 g) of Salvia divinorum, collected at Huautla, Oaxaca (Mexico) in November 1980, were extracted with boiling chloroform. Evaporation of the solvent gave a green residue ( 27 g ) which was purified by chromatography on 'Tonsil ' ( 200 g ) with chloroform as eluant. Thirteen fractions of 50.0 ml were collected, the sixth and seventh of which contained compound (1) as ascertained by t.l.c. ( $45 \%$ ethyl acetate in hexane as developer; $R_{F} 0.7$ ). Crystallization from methanol yielded salvinorin (1) as colourless crystals, m.p. 238$240^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}-41^{\circ}\left(c, 1\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; \nu_{\text {max. }} 1735 \mathrm{~cm}^{-1} ; \delta_{\mathrm{H}} 1.11$ $(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.45(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.16(3 \mathrm{H}, \mathrm{s}, \mathrm{COMe}), 3.74(3 \mathrm{H}$, $\left.\mathrm{s}, \mathrm{CO}_{2} \mathrm{Me}\right), 5.14(1 \mathrm{H}, \mathrm{t}, J 10,2-\mathrm{H}), 5.51(1 \mathrm{H}, \mathrm{dd}, J 12$ and 6 , $12-\mathrm{H}), 6.38(1 \mathrm{H}, \mathrm{m}, 14-\mathrm{H})$, and $7.41(2 \mathrm{H}, \mathrm{m}, 15-\mathrm{and} 16-\mathrm{H})$; $\delta_{\mathrm{C}} 15.19(\mathrm{q}, \mathrm{C}-19), 16.36(\mathrm{q}, \mathrm{C}-20), 18.11\left(\mathrm{t}, \mathrm{CH}_{2}\right), 20.56$ ( $\mathrm{q}, \mathrm{C}-22$ ), 30.75 ( $\mathrm{t}, \mathrm{CH}_{2}$ ), 35.41 ( $\mathrm{s}, \mathrm{C}-9$ ), 38.08 ( $\mathrm{t}, \mathrm{C}-11$ ), $42.06(\mathrm{~s}, \mathrm{C}-5), 43.23\left(\mathrm{t}, \mathrm{CH}_{2}\right), 51.26(\mathrm{~d}, \mathrm{C}-8), 51.90(\mathrm{q}, \mathrm{C}-23)$, 53.47 (d, C-4), 63.90 (d, C-10), 72.00 (d, C-12), 75.03 (d, $\mathrm{C}-2$ ), 108.41 ( $\mathrm{d}, \mathrm{C}-14$ ), 125.25 ( $\mathrm{s}, \mathrm{C}-13$ ), 139.46 ( $\mathrm{d}, \mathrm{C}-16$ ), 143.66 (d, C-15), 169.94 (s, C-21), 171.15 ( $\mathrm{s}, \mathrm{C}-18$ ), 171.57 (s, C-17), and 202.04 p.p.m. (s, C-1) (assignments are tentative and are based on chemical shifts and off-resonance decoupled spectra); $m / z 432\left(M^{+}, 20\right), 404$ (15), 359 (5), 318 (20), 273 (30), and 94 (100) (Found: C, 63.5; H, 6.3. $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{O}_{8}$ requires $\mathrm{C}, 63.88 ; \mathrm{H}, 6.53 \%$ ).

X-Ray Crystallographic Analysis of Salvinorin (1).$\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{O}_{8}, \quad M=432.47$. Orthorhombic, space group $P 2_{1} 2_{1} 2_{1}, \quad a=6.368(2), \quad b=11.338(3), \quad c=30.710(6) \quad \AA$; $Z=4 ; \quad D_{\mathrm{c}}=1.295 \mathrm{~g} \mathrm{~cm}^{-3} ; \mu\left(\mathrm{Cu}-K_{\alpha}\right)=8.3 \mathrm{~cm}^{-1}$. The intensity data, uncorrected for absorption, were measured on a fully automated Hilger-Watts diffractometer (Nifiltered $\mathrm{Cu}-K_{\alpha}$ radiation; $\theta-2 \theta$ scans; pulse-height discrimination) using a crystal of dimensions ca. $0.08 \times$ $0.20 \times 0.6 \mathrm{~mm}$ grown from methanol. Of 1763 independent reflections for $\theta<57^{\circ}, 1518$ were considered to be observed $[I>2.5 \sigma(I)]$. The structure and relative stereochemistry of compound (1) were solved by a multiplesolution procedure ${ }^{4}$ and refined by full-matrix least-squares. In the final refinement the non-hydrogen atoms were refined anisotropically, except for the oxygen atom of a molecule of water, which was refined isotropically. The occupancy factor of the oxygen atom of the water molecule was included in the refinement and was found to be $0.32(1)$. The hydrogen atoms were included in the structure-factor calculations but their parameters were not refined. The final discrepancy indices were $R 0.052, R^{\prime} 0.056$ for the 1518 observed reflections. The final difference map had no peaks greater than $0.2 \mathrm{e} \AA^{-3}$. Listings of final atomic parameters, bond lengths, bond angles, and torsion angles are given in Tables 1-4. Observed and calculated structure factors and atomic thermal parameters are given in Supplementary Publication No. SUP 23371 ( 8 pp.$) \cdot \dagger$

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## REFERENCES

${ }^{1}$ A. F. Halim and R. P. Collins, J. Agric. Food Chem., 1975, 23, 506; W. H. Lewis and M. P. F. Elvin-Lewis, 'Medical Botany,' Wiley, New York, 1977, p. 338.
${ }^{2}$ J. M. Watt in ' Plants in the Development of Modern Medicine,' ed. T. Swain, Harvard University Press, 1972, p. 67.
${ }^{3}$ G. Savona, M. P. Paternostro, F. Piozzi, J. R. Hanson, P. B. Hitchcock, and S. A. Thomas, J. Chem. Soc., Perkin Trans. $1,1978,643$, and references cited.

- G. Germain, P. Main, and M. M. Woolfson, Acta Crystallogr., Sect. A, 1971, 27, 368.
${ }_{5}$ M Martinez-Ripoll, J. Fayos, B. Rodriguez, M. Paternostro, and J. R. Hanson, J. Chem. Soc., Perkin Trans. 1, 1981, 1186.
${ }^{6}$ D. Rogers, G. G. Unal, D. J. Williams, S. V. Ley, G. A. Sim,
B. S. Joshi, and K. R. Ravindranath, J. Chem. Soc., Chem. Commun., 1979, 97; G. Trivedi, H. Komura, I. Kubo, K. Nakanishi, and B. S. Joshi, ibid., 1979, 885; I. Kubo, M. Kido, and Y. Fukuyama, ibid., 1980, 897; F. Piozzi, Heterocycles, 1981, 15, 1489.
${ }^{7}$ For reviews see J. R. Hanson in ' Terpenoids and Steroids,' Specialist Periodical Reports, The Chemical Society, London, 1981 , vol. 10 and preceding volumes.
${ }^{8}$ G. Savona, M. P. Paternostro, F. Piozzi, and J. R. Hanson,
J. Chem. Soc., Perkin Trans. 1, 1979, 533.

